





# Supramolecular microcapsules: *directing self-assembly*

Richard M. Parker, Oren A. Scherman and Chris Abell

Department of Chemistry and the Melville Laboratory for Polymer Synthesis, University of Cambridge, UK

rmp53@cam.ac.uk

http://www-microdroplets.ch.cam.ac.uk/research.html

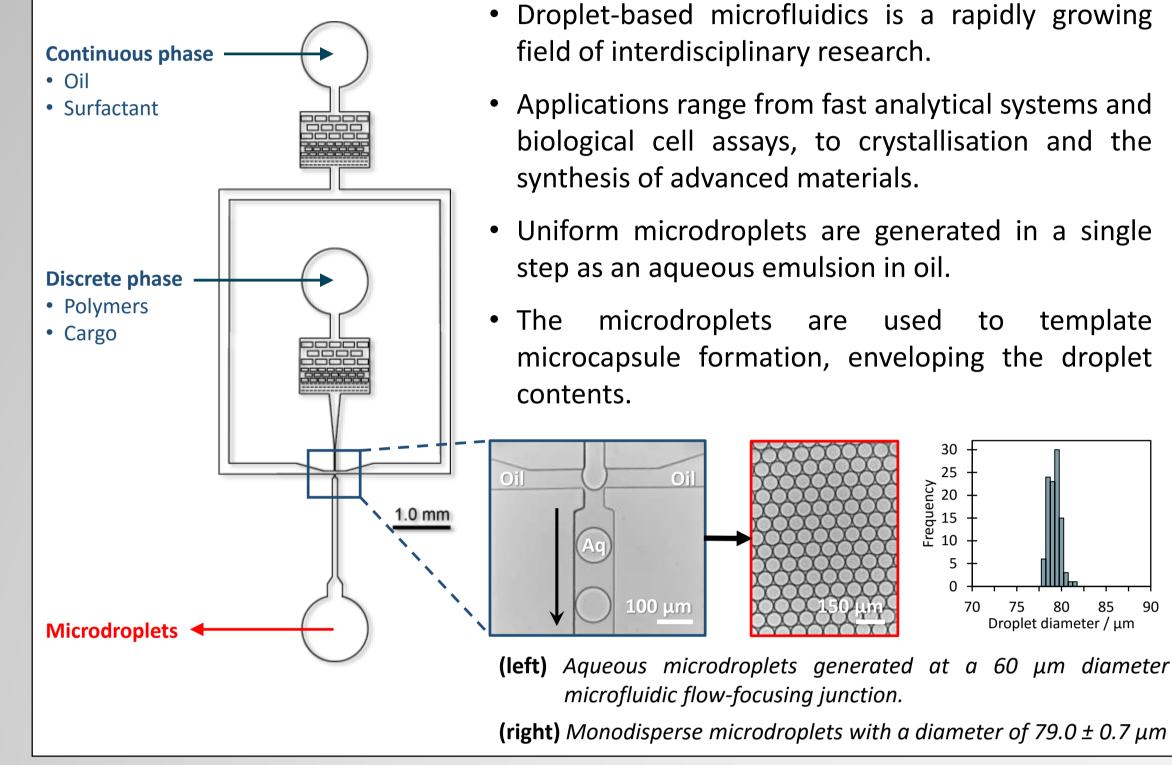
#### Supramolecular Microencapsulation

Microencapsulation refers to a very wide range of technologies that encapsulate, protect and release active cargo when needed. The annual global market is estimated at \$40 billion in 2015 with applications across a wide range of products; from detergent and perfume, to paints and pesticides. It also has potential in other areas, including: targeted drug delivery, cell encapsulation, catalysis and self-healing concrete.

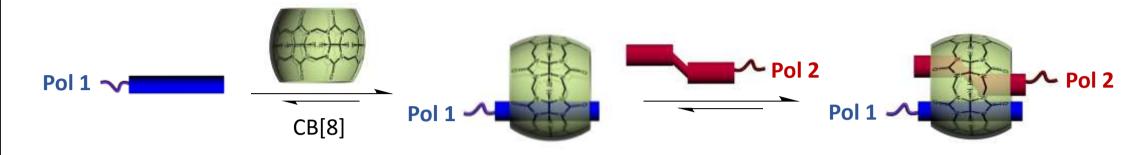


Directing both the micro-scale accumulation and molecular-scale self-assembly of components at the interface of sub-millimetre aqueous 'microdroplets' offers a powerful route to monodisperse 'microcapsules' with identical composition, in a single step. These microcapsules are uniquely assembled by dynamic molecular "handcuffs" that can be triggered to dismantle when exposed to a specific stimulus (e.g. light), releasing the protected cargo on demand.

# **Droplet-based microfluidics**



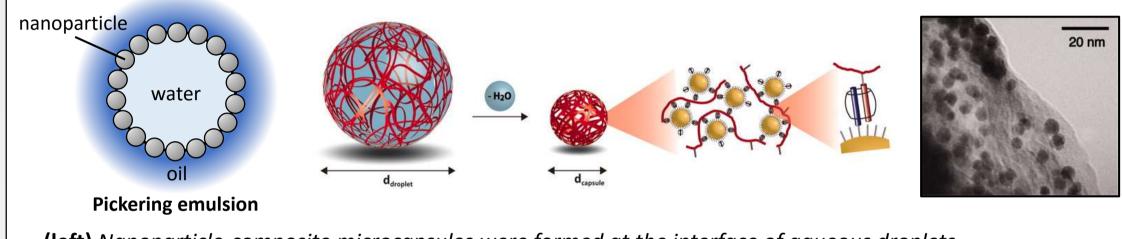
## Ternary Host-Guest complex



Supramolecular microcapsules are assembled *via* a dynamic ternary complex between cucurbit[8]uril (CB[8]) and guest molecules on the substrates.

#### Nanoparticle-driven assembly <sup>[1,2]</sup>

Supramolecular *nano-composite* microcapsules and colloidosomes self-assemble by first forming a 'Pickering emulsion' of nanoparticles at the oil/water droplet interface.



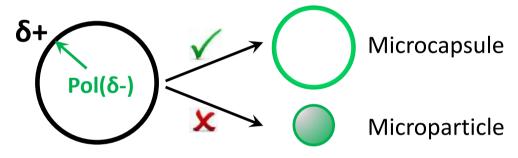
(left) Nanoparticle-composite microcapsules were formed at the interface of aqueous droplets. (right) Image of the capsule skin, showing gold nanoparticles embedded within a polymer mesh.

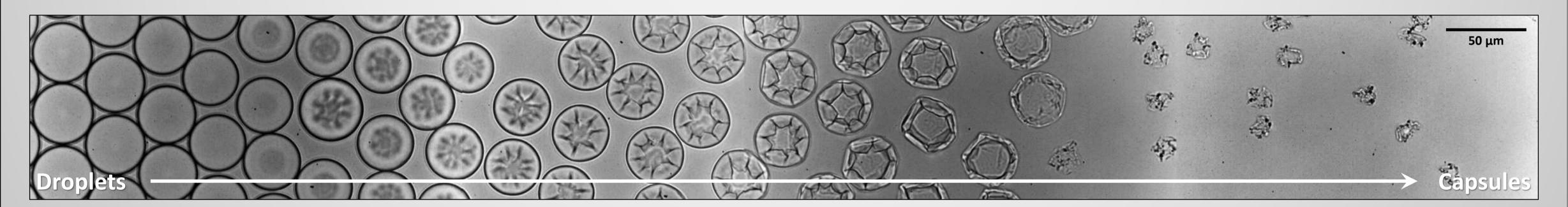
#### **Polymer-only microcapsules**

To expand the versatility of the supramolecular platform it is necessary to generalise microcapsule fabrication away from the need to incorporate nanoparticles.

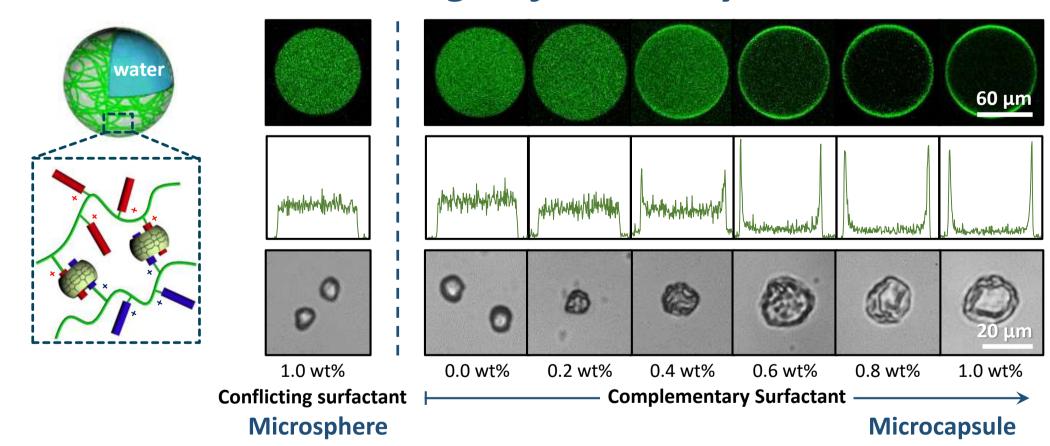
The assembly of *supramolecular microcapsules* from *aqueous microdroplets* is driven by electrostatic interactions, whereby charged polymers are selectively accumulated at the microdroplet interface by a complementary-charged surfactant (patent filed and licensed).

This is both dynamic and reversible, with the location of polymers within the droplet able to be externally manipulated through the carrier oil.

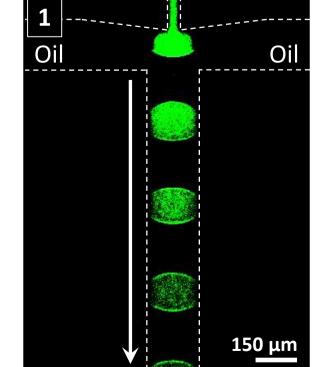




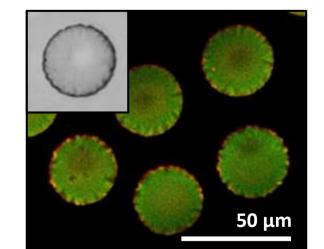
#### **Directing self-assembly**<sup>[4]</sup>



(top) Increasing the amount of complementary-charged surfactant results in an accumulation of polymer at the interface, leading to a switch from microparticle to microcapsule formation.



- The distribution of polymer within the microdroplet directly correlates with the resultant micro-structure upon evaporation.
- *Conflicting charge:* microdroplets contain a uniform distribution of polymer throughout the droplet, forming a smooth *microparticle*.
- **Complementary charge:** partitioning of all polymers to the droplet interface leads to the formation of an ultrathin,



### Core-shell capsules [4]

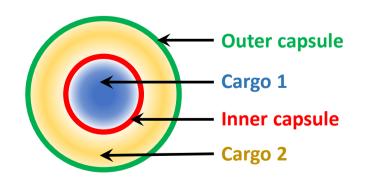
- Control over the location of individual components within the droplet allows for the design of complex structures.
- A mixed solution of oppositely-charged polymers results in the formation of core-shell capsules (*left*):
  - red polymer forms the outer capsule wall
  - green polymer forms the solid core

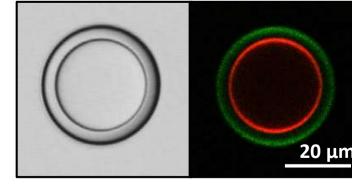
## 'Capsules-in-Capsules'

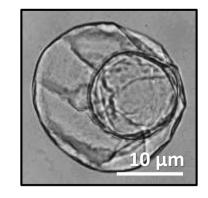
Electrostatics can be extended to multiple interfaces within a nested microdroplet to form capsules within capsules. Here orthogonal charges allow for distinct compartments.

- Capsule architecture is externally controlled.
- Unique chemistry at each interface ('bespoke').
- Precise, multi-step or multi-trigger release.
- Synergistic delivery with controlled dosages.
- Study of chemistry in a controlled environment.

• Segregated carriage of multiple cargos.

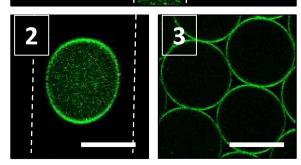






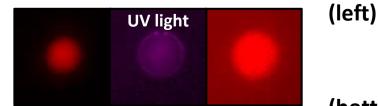
(left) Schematic of a compartmentalised 'Capsule-in-Capsule' with segregated cargo.

(right) Partitioning of a mixture of oppositely-charged polymers to specific interfaces within the nested droplet results on evaporation in the formation of two distinct capsule skins.



#### hollow *microcapsule*.

Partitioning of charged polymers to the droplet interface is both rapid (<3 s) and dynamic.



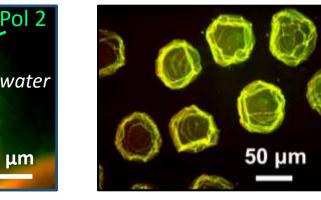
Cargo retained ——— Cargo released

Tracking the accumulation of charged polymer at the interface as the droplet flows along the microfluidic channel from **1** to **3** (~10 s), by fluorescence.

(bottom) Triggered release of capsule cargo (red) with UV light.

#### Alternative approach: Interfacial assembly <sup>[3]</sup>

Microdroplet  $\longrightarrow$  Microcapsule chloroform



Complementary polymers self-assemble at the microdroplet interface through the supramolecular "handcuff" (*left*), forming a polymer microcapsule (*right*).

#### REFERENCES

[1] J. Zhang, R.J. Coulston, S.T. Jones, J. Geng, O.A. Scherman, C. Abell, *Science*, 2012, **335**, 690–694. [2] G. Stephenson, R.M. Parker, Y. Lan, Z. Yu, O.A. Scherman, C. Abell, *Chem. Commun.*, 2014, **50**, 7048–7051. [3] Y. Zheng, Z. Yu, R.M. Parker, Y. Wu, C. Abell, O.A. Scherman, *Nat. Commun.*, 2014, 5, 5772. [4] (a) R.M. Parker et al., Adv. Funct. Mater., 2015, 25, 4091. (b) Z. Yu, et al., Chem. Sci., 2015, 6, 4929.