





Supramolecular microcapsules: *directing self-assembly*

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Supramolecular Microencapsulation

Microencapsulation refers to a very wide range of technologies that encapsulate, protect and release active cargo when needed. The annual global market is estimated at \$40 billion in 2015 with applications across a wide range of products; from detergent and perfume, to paints and pesticides. It also has potential in other areas, including: targeted drug delivery, cell encapsulation, catalysis and self-healing concrete.



Directing both the micro-scale accumulation and molecular-scale self-assembly of components at the interface of sub-millimetre aqueous 'microdroplets' offers a powerful route to monodisperse 'microcapsules' with identical composition, in a single step. These microcapsules are uniquely assembled by dynamic molecular "handcuffs" that can be triggered to dismantle when exposed to a specific stimulus (e.g. light), releasing the protected cargo on demand.

Droplet-based microfluidics



Ternary Host-Guest complex



Supramolecular microcapsules are assembled *via* a dynamic ternary complex between cucurbit[8]uril (CB[8]) and guest molecules on the substrates.

Nanoparticle-driven assembly ^[1,2]

Supramolecular *nano-composite* microcapsules and colloidosomes self-assemble by first forming a 'Pickering emulsion' of nanoparticles at the oil/water droplet interface.



(left) Nanoparticle-composite microcapsules were formed at the interface of aqueous droplets. (right) Image of the capsule skin, showing gold nanoparticles embedded within a polymer mesh.

Polymer-only microcapsules

To expand the versatility of the supramolecular platform it is necessary to generalise microcapsule fabrication away from the need to incorporate nanoparticles.

The assembly of *supramolecular microcapsules* from *aqueous microdroplets* is driven by electrostatic interactions, whereby charged polymers are selectively accumulated at the microdroplet interface by a complementary-charged surfactant (patent filed and licensed).

This is both dynamic and reversible, with the location of polymers within the droplet able to be externally manipulated through the carrier oil.





Directing self-assembly^[4]



(top) Increasing the amount of complementary-charged surfactant results in an accumulation of polymer at the interface, leading to a switch from microparticle to microcapsule formation.



- The distribution of polymer within the microdroplet directly correlates with the resultant micro-structure upon evaporation.
- *Conflicting charge:* microdroplets contain a uniform distribution of polymer throughout the droplet, forming a smooth *microparticle*.
- **Complementary charge:** partitioning of all polymers to the droplet interface leads to the formation of an ultrathin,



Core-shell capsules [4]

- Control over the location of individual components within the droplet allows for the design of complex structures.
- A mixed solution of oppositely-charged polymers results in the formation of core-shell capsules (*left*):
 - red polymer forms the outer capsule wall
 - green polymer forms the solid core

'Capsules-in-Capsules'

Electrostatics can be extended to multiple interfaces within a nested microdroplet to form capsules within capsules. Here orthogonal charges allow for distinct compartments.

- Capsule architecture is externally controlled.
- Unique chemistry at each interface ('bespoke').
- Precise, multi-step or multi-trigger release.
- Synergistic delivery with controlled dosages.
- Study of chemistry in a controlled environment.

• Segregated carriage of multiple cargos.







(left) Schematic of a compartmentalised 'Capsule-in-Capsule' with segregated cargo.

(right) Partitioning of a mixture of oppositely-charged polymers to specific interfaces within the nested droplet results on evaporation in the formation of two distinct capsule skins.



hollow *microcapsule*.

Partitioning of charged polymers to the droplet interface is both rapid (<3 s) and dynamic.



Cargo retained ——— Cargo released

Tracking the accumulation of charged polymer at the interface as the droplet flows along the microfluidic channel from **1** to **3** (~10 s), by fluorescence.

(bottom) Triggered release of capsule cargo (red) with UV light.

Alternative approach: Interfacial assembly ^[3]

Microdroplet \longrightarrow Microcapsule chloroform



Complementary polymers self-assemble at the microdroplet interface through the supramolecular "handcuff" (*left*), forming a polymer microcapsule (*right*).

REFERENCES

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